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interactive guided tutorial. with ChemASAP Figure 4.8 Au is the chemical symbol for gold. Applying Concepts How many electrons does a gold atom have? 197 111 Mass Number that most of the mass of an atom is concentrated in its nucleus depends on the number of protons and neutrons. The total number of protons and neutrons in an atom is called the mass number.

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of pressure

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birthday cake for a four-year-old has four candles.

9.5 The Laws Governing Formulas and Names

Sketch two containers with identical volumes on the board. Tell students one container is filled with neon gas and the other with helium at the same temperature and pressure. Ask students to use the ideal gas law to find the equation for the number of moles in each container. Students should determine that

14.3 Ideal Gases

of an interactive guided tutorial. For help with using a calculator, go to page R62. 18. Find the number of moles in 3.70 $\times 10^1$ g of boron. Rust weakens an iron chain. SAMPLE PROBLEM 10.6 Converting Mass to Moles When iron is exposed to air, it corrodes to form red-brown rust. Rust is iron(III) oxide (Fe_2O_3). How many moles of iron(III) oxide ...

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10.2 Mole-Mass and Mole-Volume Relationships 10

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