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1 Basic Problems of Chemical Reaction Engineering and ...

The kinetic and the potential energy are negligible in many chemical reaction engineering applications, so Equation 15 becomes $Q V E = Q V (U + K + E P) = Q V U$ (16) we know that Uis a function of the enthalpy, pressure, and volume, so $Q V U = Q V (H + PV) = Q V X N c i = 1 n i h i P Q = X N c i = 1 F i h i P Q$ (17) When this is applied for both streams, the term PQcancels with the volumetric work

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Some specific examples of chemical engineering solutions that have great potential for wider application, and thus are avenues for potential innovation, include: process intensification, i.e., any chemical engineering development that leads to a substantially smaller, cleaner, and more energy-efficient technology (6)

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Chemical reaction engineering - Wikipedia

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